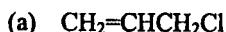
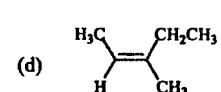
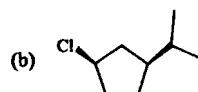
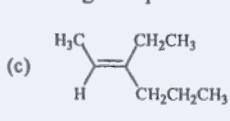
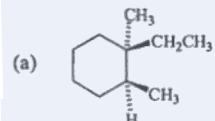


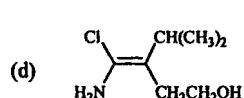
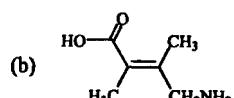
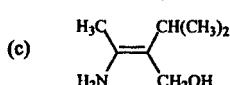
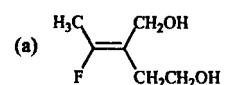
1. Write Lewis (electron-dot) structures for the following molecules: (10%)



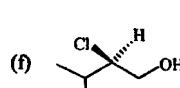
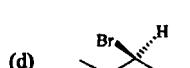
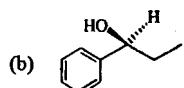
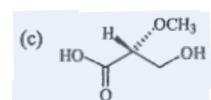
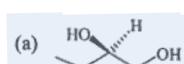
2. Give IUPAC names for the following compounds: (12%)



3. Assign E or Z configuration to the following alkenes: (8%)



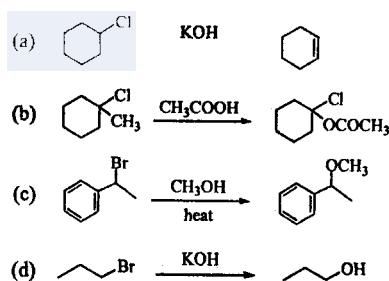
4. Assign R or S configuration to the stereocenters in the following molecules: (12%)



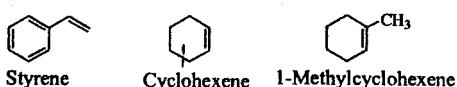
5. Show the steps involved in the Friedel-Crafts reaction of benzene with  $\text{CH}_3\text{Cl}$ .

(8%)

6. Identify the following reactions are either  $S_N1$ ,  $S_N2$ ,  $E1$ , or  $E2$ : (16%)

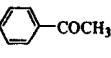


7. Predict the products of the following reactions. Indicate regioselectivity where relevant. (The aromatic ring is inert to all the indicated reagents.) (20%)



- (a) Styrene +  $\text{H}_2\text{SO}_4 + \text{H}_2\text{O}$   $\longrightarrow$
- (b) Styrene +  $\text{HBr}$   $\longrightarrow$
- (c) Styrene +  $\text{KMnO}_4 + \text{H}_3\text{O}^+$   $\longrightarrow$
- (d) Cyclohexene +  $\text{KMnO}_4 + \text{H}_3\text{O}^+$   $\longrightarrow$
- (e) Cyclohexene +  $\text{Br}_2$   $\longrightarrow$
- (f) Cyclohexene +  $\text{KMnO}_4 + \text{NaOH}, \text{H}_2\text{O}$
- (g) 1-Methylcyclohexene +  $\text{KMnO}_4 + \text{H}_3\text{O}^+$
- (h) 1-Methylcyclohexene +  $\text{H}_2\text{SO}_4 + \text{H}_2\text{O}$  -
- (i) 1-Methylcyclohexene +  $\text{Br}_2$  -
- (j) 1-Methylcyclohexene +  $\text{HCl}$   $\longrightarrow$

8. Suggest structures that give the following reaction products. (14%)

- (a) ? +  $\text{KMnO}_4 + \text{H}_3\text{O}^+$   $\longrightarrow$   $\text{CH}_3\text{COCH}_3$
- (b) ? +  $\text{KMnO}_4 + \text{H}_3\text{O}^+$   $\longrightarrow$   $\text{CO}_2 + \text{CH}_3\text{CH}_2\text{COOH}$
- (c) ? +  $\text{HBr}$   $\longrightarrow$  2-Bromo-2-methylbutane
- (d) ? +  $\text{KOH}$  in  $\text{CH}_3\text{CH}_2\text{OH}$   $\longrightarrow$  2-methyl-2-pentene
- (e) ? +  $\text{H}_2\text{SO}_4$   $\longrightarrow$  2,3-dimethyl-2-pentene
- (f) ? +  $\text{H}_2\text{O} + \text{H}_2\text{SO}_4 + \text{HgSO}_4$   $\longrightarrow$  
- (g) ? +  $\text{KMnO}_4 + \text{H}_3\text{O}^+$   $\longrightarrow$   $\text{CH}_3\text{COCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{COCH}_3$